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## Synthesis of L-Carbafuranomycin, an Unnatural Analogue of the Antibiotic Amino Acid Furanomycin<sup>†,‡</sup>

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## **ABSTRACT**

COOH 
$$\Rightarrow$$
 OR  $\Rightarrow$  ONC ONC ONC ONC ONC ONC

L-(+)-Carbafuranomycin is a novel analogue of L-(+)-furanomycin, an unusual antibiotic  $\alpha$ -amino acid that attracted great interest due to its activity as an isoleucine antagonist. We present here a concise and efficient asymmetric synthesis of this carba-analogue starting with the 1,3-dipolar cycloaddition of a chiral nitrile oxide with cyclopentadiene. Notably, the methyl group was introduced by an  $S_N2'$  cuprate substitution with high stereo- and regioselectivity.

Furanomycin 1 is a naturally occurring antibiotic  $\alpha$ -amino acid that was isolated from metabolites of *Streptomyces threomyceticus* in 1967 by Katagiri et al.<sup>1</sup> This unusual  $\alpha$ -amino acid suppresses the growth of several bacterial species such as *E. coli*, *Bacillus subtilis*, or *Shigella* and *Salmonella* strains.<sup>1</sup>

Furanomycin 1 competitively inhibits the isoleucyl aminoacyl tRNA synthetase. In vitro experiments have shown that furanomycin is incorporated in proteins instead of isoleucine.<sup>2</sup> It is further remarkable that furanomycin is a substrate for isoleucyl tRNA synthetase, considering that the structures of isoleucine and furanomycin differ considerably. NMR studies showed, however, that the conformation of enzyme-bound furanomycin is very similar to that of isoleucine.<sup>2</sup> These types of translatable amino acids are of

great interest for the preparation of peptides or proteins containing unusual, i.e., nonproteinogenic amino acids.<sup>3,4</sup>

Until now, several syntheses of **1**, both as a racemic mixture and as pure enantiomer, have been published. The starting points for most of the enantioselective syntheses were either substituted furans<sup>5</sup> or carbohydrates.<sup>6</sup> The latest reports include a Ag(I)-catalyzed cyclization of an allenic alcohol<sup>4</sup> and a 1,3-dipolar cycloaddition of a chiral nitrile oxide with methylfuran, respectively.<sup>7</sup> Apart from this, several strategies for routes to stereoisomers of **1** like 5-*epi*-furanomycin have likewise employed substituted furans<sup>6a</sup> or carbohydrate precursors.<sup>8,9</sup>

<sup>†</sup> Syntheses via Isoxazolines, Part 27. For Part 26, see ref 7b.

<sup>&</sup>lt;sup>‡</sup> Dedicated to Professor Günther Helmchen on the occasion of his 65th birthday.

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Concerning furanomycin analogues, syntheses of norfuranomycin, 10 cyclopentylglycine, 11 cyclopentenyl-glycine, 12 and cyclohexenylglycine<sup>12b,13</sup> and their biological activities have been disclosed. Norfuranomycin showed antibiotic activity against E. coli and several species of Pseudomonas strains. The other compounds also inhibited the growth of E. coli, but details were not reported in these papers. Recently, syntheses of a series of racemic furanomycin analogues such as norfuranomycin, the 3-, 4-, and  $\alpha$ -methyl regioisomers of furanomycin, cyclopentylglycine, and cyclopentenylglycine were published, based on aldol additions and ring-closing metathesis or ester-enolate Claisen rearrangements.14 According to the biological evaluation, no activities were found apart from the case of norfuranomycin, however.<sup>14</sup> In contrast to this, 1-cyclobutenylglycine was shown to substitute valine and isoleucine, but not leucine, in the translation of GFP in vitro.<sup>4b</sup>

In view of these structure—activity studies, which showed a very narrow activity profile of the natural structure, we reasoned that carbafuranomycin 2 would present only a minute and tolerable change (Figure 1).

Figure 1. L-Furanomycin and carba-analogues.

This paper describes an enantioselective synthesis of 2 where the dihydrofuran oxygen atom has been replaced by a  $CH_2$  unit, along with preliminary tests of its biological activity.

A main feature of this synthesis (Scheme 1) is the 1,3-dipolar cycloaddition of cyclopentadiene and the chiral nitrile oxide 3. On one hand, the cyclopentene moiety of carbafuranomycin 2 is supplied by cyclopentadiene; on the other hand, the chiral cyclopentaisoxazoline 4 contains two stereogenic centers in the carbocycle, from which the two others, in the ring and in the side-chain of the target structure, were to be elaborated. Thus, the stereoselective reduction of 4 would lead to the cyclopentenol derivative 5, the precursor for the crucial step, the introduction of the methyl group by an  $S_{\rm N}2'$  reaction.

Scheme 1. Retrosynthesis of Carbafuranomycin 2

COOH 
$$\longrightarrow$$
  $\bigcap_{NH_2} R^*$   $\longrightarrow$   $\bigcap_{O-N} R^*$   $\longrightarrow$   $\longrightarrow$   $+$   $\begin{bmatrix} - & + \\ O-N & - \\ & & \end{bmatrix}$ 

The required nitrile oxide precursor, the hydroximoyl chloride  $\mathbf{6}$ , was obtained according to earlier work<sup>15,16</sup> from the mannitol bis(acetonide)  $\mathbf{7}$  by periodate cleavage, oximation, and subsequent chlorination with N-chloro-succinimide (NCS)<sup>16b</sup> in 75% overall yield (Scheme 2).

**Scheme 2.** Synthesis of the Hydroximoyl Chloride **6** 

The nitrile oxide was released in situ in low concentration by slow addition of triethylamine to **6**, in order to minimize furoxan formation (dimerization). In the presence of excess, freshly distilled cyclopentadiene, the dipole underwent cycloaddition to provide a diastereomeric mixture of **8** and **9** in 80% yield (dr 56:44).

Separation of **8** and **9** proved to be difficult, so diol protection was switched from the acetonide to benzyl in **10** and **11**, which were separable by MPLC. Thus, the mixture of **8** and **9** was treated with dilute trifluoroacetic acid in methanol and water (4:1) to give the diols, which were benzylated with benzyl bromide and sodium hydride in DMF in 78% yield for the two steps. At this stage, the configuration of **10** and **11** was assigned as (3aS,6aS,1'S)- ("anti") and (3aR,6aR,1'S)- ("syn"), respectively, on the basis of the specific rotations amounting to  $[\alpha]_0^{20} = -36.9$  and +135.3.17

Next, the stereoselective reduction of the isoxazoline to the  $\beta$ -amino alcohol was envisaged, a reliable transformation well known from earlier work. The cyclopentaisoxazoline 10 was reduced to the amino alcohol by action of lithium aluminum hydride in diethyl ether. Due to the bicyclic, bowlshaped structure, hydride attack occurred preferentially from the sterically less hindered *exo*-side, as expected. The free

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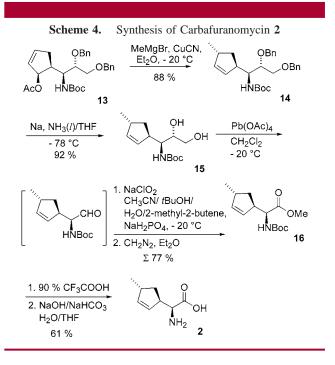
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amino group was then protected with  $Boc_2O$  in dioxane and water (3:1) to give the cyclopentenol **12**, plus a minor product (probably a diastereoisomer, ratio = 90:10), which was separated by MPLC (87% yield of **12** from **10**). To act as a leaving group, the hydroxy group was acetylated with acetic anhydride in pyridine with a catalytic amount of DMAP to provide the cyclopentenyl acetate **13** in 97% yield (Scheme 3).

**Scheme 3.** Synthesis of the Cyclopentenyl Acetate **13** Starting with 1,3-Dipolar Nitrile Oxide Cycloaddition

For the next step, the key introduction of the methyl group, use of an organocuprate seemed promising. Organocuprates  $^{19}$  are known to react with suitable cyclic systems in an  $S_{\rm N}2'$  fashion, i.e., nucleophilic attack at the position  $\gamma$  to the leaving goup, with relocation of the double bond.  $^{20}$  Furthermore, anti attack is stereochemically favored.  $^{20}$  On this basis, we assumed that the required methylcyclopentenyl moiety might be obtained with the correct regio- and stereochemical preference.

The acetate **13** was added dropwise to the methyl cuprate,<sup>21</sup> prepared from 5 equiv of CuCN and 10 equiv of MeMgBr in diethyl ether, and provided compound **14** in 88% yield with a ratio of 88:12 (Scheme 4). The minor product,



probably the regioisomer from the competing  $S_N2$  reaction, was not identified. The isomer ratio was improved from 88: 12 to >95:5 by recrystallizing the diol **15** (vide infra). For this substitution, 5 equiv of the "higher order" cyanocuprate were required, probably because of the additional functional groups in **13** with complexing ability. Experiments with less cuprate proceeded too slowly or with incomplete reaction.

For the next step, twofold O-debenzylation, catalytic hydrogenation was presumed to be incompatible with the

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<sup>(17) (</sup>a) Such large, consistent differences in optical rotation had been noted earlier with each one of some 60 5- and 4,5-cis-substituted isoxazolines whose absolute configuration had been established by chemical correlation and/or crystal structure analyses; see refs 7, 16a, 18b, and 18c. For example, in the furanomycin series, the respective furoisoxazolines (methylfuran adducts of the nitrile oxide generated from 6 in a 60:40 ratio) showed  $[\alpha]_D^{20}=-171$  for the (3aS,6aS,1'S)-isomer ("anti"), corresponding to 10, and  $[\alpha]_D^{20}=+199$  for the (3aR,6aR,1'S)-diastereomer ("syn"), corresponding to 11; see ref 7. (b) Müller, R. Dissertation, Universität Würzburg, Würzburg, Germany, 1992. (c) Leibold, T. Dissertation, Universität Stuttgart, Stuttgart, Germany, 1995.

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<sup>(21)</sup> To a solution of 5 equiv of CuCN in diethyl ether at -20 °C was added 10 equiv of CH<sub>3</sub>MgBr in diethyl ether (3.0 M). After 20 min at -20 °C, the acetate **13** in diethyl ether was added slowly and kept for 1.5 h at -20 °C.

C-C double bond present; therefore, sodium metal in liquid ammonia was used. Thus, at -78 °C and short exposure,  $^{22}$ the benzyl ether was transformed to the diol 15 in 92% yield. The diol moiety served as a latent carboxyl group as had been exploited earlier.<sup>23</sup> The diol 15 was cleaved by lead tetraacetate in CH<sub>2</sub>Cl<sub>2</sub> to afford the intermediate aldehyde, which was directly oxidized to the carboxylic acid by means of sodium chlorite (NaClO<sub>2</sub>) in a phosphate buffer.<sup>24</sup> Despite these very weakly acidic conditions, some epimerization was observed (probably at the  $\alpha$ -amino aldehyde stage), which required immediate conversion to the diastereomerically pure ester 16 using diazomethane (77% yield over three steps). Finally, the Boc group was removed with 90% trifluoroacetic acid, and the ester was hydrolyzed under mildly basic conditions (pH 10.5). After ion-exchange column purification (Dowex 50WX8, H<sup>+</sup> form), L-carbafuranomycin 2 was obtained in analytically and spectroscopically pure form in 61% yield from the *N*-Boc methyl ester **16**.

The structure of **2**, with the methyl substituent at C5′, and its configuration were confirmed notably by  $^1H$  NMR coupling constants and chemical shifts. The chemical shift difference for 4′- $H_a$  and 4′- $H_b$  in **2** was a mere 0.26 ppm, characteristic for trans-3,5-disubstituted cyclopentenes.  $^{20k,25}$ 

The optical rotation of **2** was found to be  $[\alpha]_D^{20} = +164$  (c = 0.08,  $H_2O$ ), mp 215-217 °C. Antibacterial tests with 2 were performed with several bacterial species such as E. coli. S. aureus, and B. subtilis; no significant activity was detected, however.<sup>26,27</sup> The minimal inhibitory concentration (MIC) of 2 was > 100  $\mu$ M, with the exception of the efflux pumpdeficient E. coli HN818 (MIC = 25  $\mu$ M (12.5  $\mu$ M in the presence of membrane permeabilizer polymycin B nonapeptide)).<sup>27</sup> This again confirms the narrow structure—activity relationship of furanomycin.<sup>27</sup> In summary, an efficient synthesis of the new carba-analogue of L-furanomycin is presented, featuring 1,3-dipolar nitrile oxide cycloaddition and S<sub>N</sub>2' organocuprate substitution as key steps. This strategy also would give access to the enantiomer of 2 using the minor cycloadduct 11 and to the respective cyclopentenylglycines.

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**Supporting Information Available:** Detailed experimental procedures, spectral data (<sup>1</sup>H and <sup>13</sup>C NMR, IR, elemental analysis, and MS), <sup>1</sup>H NMR spectra of **10**, **12**–**16**, and **2**, and <sup>13</sup>C NMR of **10**, **13**–**15**, and **2**. This material is available free of charge via the Internet at http://pubs.acs.org.

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<sup>(22)</sup> **14** in THF was added to liquid ammonia at -78 °C, and then 10 equiv of Na was added in small pieces. The reaction was quenched after a permanent blue color appeared (ca. 5 s later). This kind of careful handling was necessary to avoid coreduction of Boc-group, observed on a small scale (<0.1 mmol).

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